



Gasoline has a specific heat of about  $2200 \text{ J/kg}\cdot\text{K}$ .

Water (as a liquid) has a specific heat of about  $4200 \text{ J/kg}\cdot\text{K}$ .

If you mix equal parts of gasoline at  $40^\circ \text{ C}$  and water at  $20^\circ \text{ C}$  the final equilibrium temperature will be:

1. below  $30^\circ \text{ C}$
2. exactly  $30^\circ \text{ C}$
3. above  $30^\circ \text{ C}$
4. dangerous to determine